

Chapter 9 Chemical Names And Formulas Worksheet Answer Key

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Chapter 9 Chemical Names And

Naming Ions & Counting Atoms: Chapter 9.1 Naming & Writing Formulas for Ionic Compounds: Chapter 9.2 Naming & Writing Formulas for Molecular Compounds: Chapter 9.3 Naming & Writing Formulas for Acids: Chapter 9.4 Quizlet Chapter 9 vocabulary Answer Keys Ionic Compound Naming and Formula Examples from notes

Chapter 9: Chemical Names & Formulas

To name a binary molecular compound, use the following guidelines: 1. Write the names of the elements in the order listed in the formula 2. Use prefixes appropriately to indicate the number of each kind of atom 3. End the name of the second element with the suffix -ide

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Compound composed of two elements and can be either ionic or molecular. First, verify that the compound is composed of a monatomic metallic cation and a monatomic nonmetallic anion. Place the cation name first, followed by the anion name. Place the correct Roman numeral after the metallic ion to indicate the charge.

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Chapter 4 "Chemical Names and Formulas" Modified and adapted from . Pre-AP Chemistry, Charles Page High School. Stephen L. Cotton, H. 2. O

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When the name of the anion ends in -ide, the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid. Naming Common Acids (2) With the anion name ends in -ite, the acid name is the stem of the anion with the suffix -ous, followed by the word acid.

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1. when the name of the anion (X) ends in -ide the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid) 2. when the anion name ends in -ite the acid name is the stem of the anion with the suffix -ous followed by the word acid (ex.

CHAPTER 9 Chemical Names And Formulas Test Review ...

Section 9.1 Regular Metals Review. The regular metals include Group 1 (except H) and 2 and aluminum. When naming a compound that starts with a regular metal, name the first element and add -ide to the second element (except for polyatomic ions).

Chapter 9 - Chemical Names and Formulas

9.2 Naming and Writing Formulas for Ionic Compounds. • The name of a binary ionic compound is the cation name followed by the anion name. • To write the formula for a binary ionic compound, write the symbol for the cation and then the anion. Then balance the charges.

CHAPTER 9 Study Guide - Ouis

Chapter 9 Chemical Names and Formulas83 SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268-270) This section explains the rules for naming and writing formulas for binary molecular compounds. Naming Binary Molecular Compounds (pages 268-269) 1. Circle the letter of the type(s) of elements that form binary molecular compounds.

Name Date Class CHEMICAL NAMES AND FORMULAS 9

Essential Understanding Acids and bases are ionic compounds. Their names and formulas reflect the number of hydrogen ions or hydroxide ions they contain. Lesson Summary Names and Formulas of Acids Names and formulas of acids are based on an acid's consisting of an anion and enough hydrogen ions to make the acid electrically neutral.

Chemical Names and Formulas - AP Biology

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Chapter 9 - Chemical Names and Formulas - 9 Assessment ...

Chapter 9: Chemical Names and Formulas Flashcards | Quizlet Chapter 9: Chemical Names and Formulas. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid) 2. when the anion name ends in -ite the acid name is the stem of the anion with the suffix -ous followed by the word acid (ex. H2SO3 -- X=suffite = sulfurous acid) 3.

Chapter 9 Chemical Names And Formulas Study Guide Answers

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Quizlet 7 Oct 2010 CHAPTER 9, Chemical Names and Formulas (continued). 8. Use the periodic table to write the name and formula (including charge) for: true luck for people who looks for chapter 9 chemical names formulas worksheet answer chemistry test chemical formulas and chemical compounds and like.

Chapter 9 chemical names and formulas tes...

Name Date Class CHEMICAL NAMES AND FORMULAS 9. Chapter 9 Chemical Names and Formulas85 Names and Formulas for Bases (page 273) 6. A base is a compound that produces ____ when dissolved in water. 7. How are bases named?

Chapter 9 Chemical Names And Formulas Test B Answers

Name CHEMICAL NAMES AND FO Practice Problems In your notebook, solve thefollowing problems SECTION 9.1 NAMING IONS What is the charge on the ion typically formed by: each element? a. oxygen b. iodine c. sodium e. nickel, 2 electrons lost d. aluminum f. magnesium 2.

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Unformatted text preview: Chemistry B - Chapter 9 TEST "Chemical Names & Formulas" Matching Match each item with the correct statement below. a. monatomic ion f. cation b. acid age binary compound c. base h. anion d. law of definite proportions Kie" polyatomic ion e.

Test on Chemical Names and Formulas - Chemistry B Chapter 9...

Quantitative Chemical Analysis BC Science 10 - Unit 2 - Chemical Reactions and Radioactivity, Text Page 162. Chapter 4 Graphics Package - text graphics from chapter 4. Each of the image files has the text page number in the file name, so it should be easy to find and identify the image that you need. Mr. Murray's Science Website: IPC Worksheets