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chemical formula subscript of each element reveals number of atoms contained in a single molecule of a certain compound
monoatomic ions ions formed from a single atom; remember to change nonmetals to -ide ending oxyanions polyatomic ions that contain oxygen binary ionic compounds compounds of 2 elements between metal (positive, s-block and p-block) and

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nonmetal ...

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Test A, continued ____19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. ____20. In the figure on the previous page, the volume of titration standard

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11. fourth 12. transition elements 13. 32 14. valence electrons
15. electron affinity 16. electronegativity 17. ionization energy
18. $3s^2 3p^4$ 19. atomic radius 20 ...

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Answers:

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

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