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SAT Subject Test in Chemistry 3 9. Difficulty: 2 Choice (E) is correct. The element with the lowest atomic number that has any electrons in an f orbital is cerium (atomic number 58), which has the electron configuration [Xe] 6s2 4f 1 5d1. All elements with an atomic number equal to or greater than 58 have ground-state atoms with electrons in f...

SAT Subject Tests Answer Explanations Chemistry

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